

1-мансирма

1.  $(80 : 20\% = 4) \text{H}_2\text{O} = 29 \text{ г / моль}$

2. амшак = 80

$80 + 29 = 109 \text{ электрон.}$

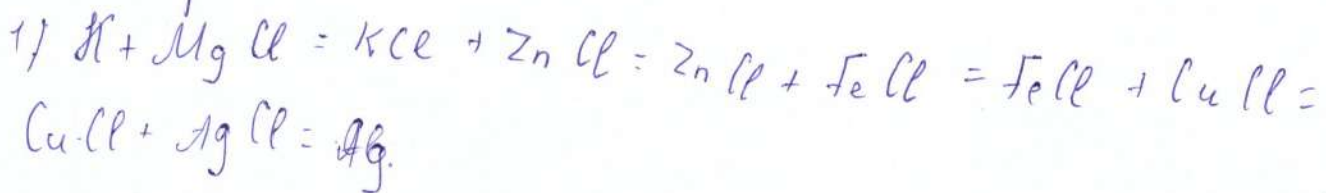
2-мансирма

1)  $M_r(\text{C}_2\text{H}_4\text{O}_2) = 12 \cdot 2 + 1 \cdot 4 + 16 \cdot 2 = 60$

2)  $G_{\text{mol}} = \frac{C_2\text{H}_4\text{O}_2}{n(\text{C}_2\text{H}_4\text{O}_2)}$

3)  $N = \frac{\text{моль}}{\text{Дм}^3/\text{моль}}$

3-мансирма



4-мансирма



1)  $\frac{2Ar(\text{Me})}{2Ar(\text{Me}) \cdot 3,82 \cdot 10^{-24}} = \frac{2Ar(\text{Me})}{3,8 \cdot 10^{-24}} + \text{HNO}_3 = \text{Ar}(\text{Me}) + \text{HNO}_3 = \text{AgNO}_3$

2)  $\text{B} = 0,4416 \text{ грамм}$       $\text{B} = \text{Cl}$

3)  $\text{B} = 35$

4)  $S = 40,506\%$       $O = 30,38\%$

